#### PHYSICAL AND CHEMICAL VARIABLES FWC/FWA **Protocols**

Version 1.1 (Dissolved Organic Carbon (DOC) added, Sep 2005)

### Aim Collection of samples from standing and running waters (or automatic recording where appropriate) for the measurement of environmentally important physical variables and for the analysis of cations and anions.

Rationale The justification for the variables selected for measurement in this ECN protocol for freshwater physical and chemical variables is provided in the Introduction to this volume under the heading "Selection of Variables" (see page 13). In general the variables have been selected because they are expected to indicate the possible causes and consequences of environmental change in the aquatic environment. Aquatic systems can be considered as consisting of a series of 'master variables' common to all freshwaters, changes in which may significantly affect the system as a whole. Other variables are mainly measures of chemical concentrations which are susceptible to changing inputs and to biogeochemical processes; in addition there are biological components of the systems which are dealt with in other protocols.

#### Method Variables to be measured and frequency of measurement

A list of the selected variables to be measured at running and standing water sites is provided in Appendix I of this protocol (page 58), together with recommended and minimum frequencies of measurement. Although there are a few exceptions, most of the variables are common to both running and standing water sites but their frequency of measurement is usually greater at the former. It is anticipated that frequency of sampling will, in practice, commonly exceed the specified minimum frequency. Detection limits for the variables are specified in Chapter 3 (pages 98-99). Water level should be recorded at the same time as the sample is taken for chemical analysis.

### Sampling – manual (FWC)

Most variables will be determined from analysis of water samples collected and handled by the methods recommended in the ECN protocol (page 49) 'Recommended sampling procedures for water chemistry' which is intended as a Guide to Good Practice. The document deals with containers, their preparation. use, labelling and storage for both running and standing waters in relation to general chemistry, trace metals, trace organics and pesticides. A widely-accepted text is referenced in the FSP protocol as a source document for the principles involved.

Should operational constraints cause any deviations from the procedures recommended in the ECN protocol, sites must send details of such deviations to the CCU.

Where manual methods are used in the field for physical determinations, (ie pH, temperature, conductivity, turbidity, dissolved oxygen), it is essential that procedures are in place for regular calibration of instruments and quality assurance.

### Sampling - automatic (FWA)

A number of both commercially available and research instruments may be used at ECN sites for measuring and recording physical variables in both running and standing waters. Variables commonly measured are pH, temperature, conductivity and turbidity. Instruments usually measure at frequent intervals and record summary values at less frequent intervals; hourly summary values will be reported for ECN purposes. Specifications for sensors and their recording parameters should be discussed with the ECN Central Co-ordination Unit before installation.

Instruments will be located centrally in standing waters and at convenient, agreed locations in running waters to coincide as far as possible with other ECN measurements.

### Procedures for chemical analysis

Laboratories of organisations participating in ECN already have well-established protocols for chemical analysis which are subject to internal quality control (QC) procedures and are accredited to formal quality assurance (QA) schemes or participate in internal or inter-laboratory checks on quality. It is therefore impractical and unnecessary to prescribe either the methods or instruments to be used for the chemical analysis of water samples. Where laboratories of organisations which intend to participate in ECN can demonstrate similar or equivalent well-established protocols and QC/QA procedures, these will be acceptable. However it is important that participating laboratories report to the CCU the details of intended methods and associated specifications for each determinand.

Author R. Owen



|   |            | Recording               | frequency <sup>(1)</sup>       | . (2)    |
|---|------------|-------------------------|--------------------------------|----------|
| Variable                                      | Running    | g waters <sup>(2)</sup> | Standing waters <sup>(2)</sup> |          |
| Water level                                   | W          | M                       | F                              |          |
| pH <sup>(3)</sup>                             | н          | M                       | H                              | õ        |
| Suspended solids <sup>(4)</sup> :             |            |                         |                                | -        |
| Drv weight                                    | W          | М                       | F                              | Q        |
| Ash-free dry weight                           | Ŵ          | M                       | F                              | Q        |
| or Turbidity <sup>(3)</sup>                   | Н          | М                       | Н                              | Q        |
| or Secchi disk                                | n/a        | n/a                     | F                              | Q        |
| Temperature <sup>(3,5)</sup>                  | Н          | M                       | H                              | Q        |
| Conductivity <sup>(3)</sup>                   | Н          | М                       | Н                              | Q        |
| Dissolved oxygen <sup>(5)</sup>               | W          | М                       | F                              | Q        |
| Ammonium: NH₄–N                               | Ŵ          | M                       | F                              | Q        |
| Total nitrogen <sup>(6)</sup>                 | Ŵ          | M                       | F                              | Q        |
| Nitrate: NO <sub>2</sub> –N                   | Ŵ          | M                       | F                              | õ        |
| Nitrite: NO <sub>2</sub> –N                   | Ŵ          | M                       | F                              | õ        |
| Alkalinity (CaCO <sub>3</sub> )               | Ŵ          | M                       | F                              | õ        |
| Chloride                                      | Ŵ          | M                       | F                              | Õ        |
| Total Organic carbon <sup>(6)</sup>           | n/a        | n/a                     | F                              | õ        |
| Dissolved Organic carbon                      | W          | M                       | F                              | õ        |
| Particulate Organic Carbon                    | n/a        | n/a                     | F                              | õ        |
| Biological Oxygen Demand                      | W          | M                       | n/a                            | n/a      |
| Total phosphorus <sup>(6)</sup>               | Ŵ          | M                       | F                              | D<br>D   |
| Particulate phosphorus                        | Ŵ          | M                       | F                              | Õ        |
| Phosphate (soluble                            | •••        | 101                     | I                              | Q        |
| reactive): PO <sub>4</sub> –P                 |            |                         |                                |          |
| Silicate: $SiO_2$                             | \٨/        | М                       | F                              | 0        |
| Sulphate: SO                                  | Ŵ          | M                       | F                              | Õ        |
| Sodium – dissolved                            | Ŵ          | M                       | F                              | Õ        |
| Sodium – total <sup><math>(6)</math></sup>    | \\/        | M                       | F                              | Õ        |
| Potassium – dissolved                         | \\/        | M                       | F                              | Õ        |
| Potassium – total <sup><math>(6)</math></sup> | Ŵ          | M                       | F                              | Õ        |
| Calcium – dissolvod                           | VV<br>\\/  | N                       | -<br>-                         | Q        |
| Calcium – total <sup>(6)</sup>                | VV<br>\\/  | M                       | F                              | Q        |
| Magnosium – dissolvod                         | VV<br>\\/  | IVI<br>M                | L L                            | Q        |
| Magnasium total <sup>(6)</sup>                | VV<br>\\/  | IVI<br>NA               | Г<br>Г                         | Q        |
| Aluminium labila(7)                           |            | IVI<br>M                | F                              | Q        |
| Aluminium – total <sup><math>(6)</math></sup> | VV<br>\\/  | IVI<br>M                | L L                            | Q        |
| Tin diagolyad                                 | VV<br>\\\/ | IVI<br>NA               | Г<br>р/р                       |          |
| Tin total <sup>(6)</sup>                      |            | IVI<br>M                | n/a                            | n/a      |
| Manganasa dissalvad                           |            | IVI<br>M                |                                | 11/a     |
| Manganaga total <sup>(6)</sup>                | VV<br>\\\/ | IVI<br>NA               |                                | Q        |
| Inanganese – total                            |            | IVI<br>NA               | F                              | Q        |
| 1001 - dissolved                              | VV<br>VV   |                         | F                              | Q        |
|   | VV         | IVI                     | F<br>n/n                       | Q<br>r/r |
| Vanadium – dissolved                          |            | IVI<br>NA               | n/a                            | n/a      |
|   | VV<br>VV   |                         | n/a                            | n/a      |
|   | VV         | IVI                     | n/a                            | n/a      |
| NICKEI – total                                | VV         | IVI<br>NA               | n/a                            | n/a      |
| Mercury – dissolved                           | VV         | IVI<br>NA               | n/a                            | n/a      |
|   | VV         |                         | n/a                            | n/a      |
| Copper – aissoived                            | VV         | IVI                     | n/a                            | n/a      |
|   | VV         | IVI                     | n/a                            | n/a      |
|   | VV         | IVI                     | n/a                            | n/a      |
|   | W          | M                       | n/a                            | n/a      |
| Cadmium – dissolved                           | W          | M                       | n/a                            | n/a      |
| Cadmium – total <sup>(*)</sup>                | W          | M                       | n/a                            | n/a      |
| Lead – dissolved                              | W          | Μ                       | F                              | Q        |
| Lead – total <sup>(9)</sup>                   | W          | Μ                       | F                              | Q        |

## **APPENDIX I.** Physical and chemical variables

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FWC/FWA Protocols

| Arsenic – total <sup>(6)</sup>  | W   | М  | F   | Q                              |
|---|---|--|---|--------------------------------|
| <b>Notes</b><br>1. Recording frequency abbrev   | viations: H<br>W<br>F<br>M<br>Q<br>n/           | hourly<br>'weekly<br>fortnightly<br>monthly<br>quarterly<br>a not applic | able  |                                |
| <ol> <li>'Standing water' and 'runnin<br/>cover more local usage suc<br/>beck, burn, stream, respecti<br/>are used interchangeably</li> </ol> | g water' are<br>h as broad, la<br>ively. Occasi | used in the tex<br>ake, loch, loug<br>ionally, the ger                   | t as generic<br>h, and river,<br>heric and loca | terms to<br>brook,<br>al terms |
| <ol> <li>Automatic (continuous) mor<br/>conductivity should be imple<br/>measured (or compensated<br/>compensated for) at 20°C.</li> </ol>    | hitoring of pH<br>mented whe<br>for) at 25°C.   | , temperature,<br>re possible. C<br>pH should be                         | turbidity and<br>onductivity s<br>measured (o   | hould be<br>or                 |
| <ol> <li>Suspended solids: dry weight<br/>free dry weight should be determined</li> </ol>   | ht should be<br>etermined at                    | determined at<br>500°±20°C.  | 105° ±5°C ar                                    | nd ash-                        |
| <ol> <li>Standing water profiles: tem<br/>monitored at depths approp<br/>depths reported to the ECN</li> </ol>                                | perature and<br>riate to the st<br>Data Manag   | l dissolved oxy<br>anding water s<br>er.                                 | gen should b<br>site, and the s                 | be<br>selected                 |
| 6. 'Total' implies analysis of the  | e <u>un</u> filtered s                          | ample.   |   |                                |
| <ol> <li>Labile aluminium should onl<br/>of pH ever having been less</li> </ol>   | y be measure<br>than 5.5.                       | ed at sites whe  | re there is a                                   | record                         |
|   |   |  |   |                                |

FWC/FWA Protocols

## Specification of results and recording conventions

The measurement variables listed below are those required for each FWC or FWA sampling location at an ECN Site. Sites submitting data to the ECNCCU should refer to the accompanying Data Transfer documentation for the specification of ECN dataset formats, available on the restricted access Site Managers' extranet. Contact <u>ecnccu@ceh.ac.uk</u> if you need access to this documentation.

The first 4 key parameters uniquely identify a sample or recording occasion in space and time, and must be included within all datasets:

Unique code for each ECN Site

- <u>Site Identification Code</u> (e.g. R10)
- Core Measurement Code (e.g. FWC)
   Location Code (e.g. 01)
   Location Code (e.g. 01)
   Each ECN Site allocates its own code to replicate sampling locations for each core measurement (e.g. FWC 01, FWC 02 for different surface water collection points)
   Sampling Date (/time)
   Date on which sample was collected or data recorded. This will include a time element where sampling is more frequent than daily

ECNCCU 2001

# Core measurement: Freshwater physical and chemical measurements – manual sampling (FWC protocol)

The following variables are recorded at a recommended frequency of weekly for rivers and fortnightly for lakes.

| Measurement<br>Variable                 | Determinand code | Units               | Reporting precision | Recommended<br>limit of<br>detection <sup>(2)</sup> |
|---|------------------|---------------------|---------------------|---|
| Site Identification Code                |                  |                     |                     |   |
| Core Measurement Code                   |                  |                     |                     |   |
| Location Code                           |                  |                     |                     |   |
| Sampling Date                           |                  |                     |                     |   |
| Sampling Time                           |                  | GMT (24-h)          |                     |   |
| Water level                             |                  | m                   | 0.001               |   |
| pH <sup>(3)</sup> level                 | PH               | pH scale            | 0.01                |   |
| Suspended solids <sup>(4)</sup>         |                  | 4                   |                     | 4   |
| Dry weight:                             | SUSS             | mg l 1              | 3 sig. figs.        | 2 mg l  |
| Ash-free dry weight:                    | SUSSAF           | mg l <sup>-</sup> ' | 3 sig. figs.        | 2 mg l <sup>-</sup> '                               |
| Turbidity <sup>(3)</sup>                | TURB             | NTU                 | 0.01                |   |
| <sup>(L)</sup> Secchi disk              | SECCI            | m                   | 0.1                 |   |
| Temperature <sup>(6)</sup>              | TEMP             | °C                  | 0.1                 | 4   |
| Conductivity <sup>(7)</sup>             | CONDY            | µS cm⁻'             | 3 sig. figs.        | 1 µS cm ัˈ  |
| Dissolved Oxygen <sup>(6)</sup>         | DISOX            | mg l 1              | 3 sig. figs.        | 0.1 mg l <sup>-1</sup>                              |
| Ammonium: NH <sub>4</sub> –N            | NH4N             | mg l 🖞              | 3 sig. figs.        | 0.01 mg l 1   |
| Total Nitrogen                          | NTOT             | mg l '              | 3 sig. figs.        | 0.01 mg l '   |
| Nitrate: NO <sub>3</sub> –N             | NO3N             | mg l 1              | 3 sig. figs.        | 0.01 mg l 1   |
| Nitrite: NO <sub>2</sub> –N             | NO2N             | mg l ً¦             | 3 sig. figs.        | 0.01 mg l 1   |
| Alkalinity (CaCO <sub>3</sub> )         | ALKY             | mg l '              | 3 sig. figs.        | 0.02 mg l 1   |
| Çhloride                                | CL               | mg l 🖞              | 3 sig. figs.        | 0.05 mg ∫ <sup>⁻</sup> ′                            |
| <sup>(L)</sup> Total Organic Carbon     | СТОТ             | mg l ً¦             | 3 sig. figs.        | 0.1 mg l 1  |
| Dissolved Organic Carbon                | DOC              | mg l '              | 3 sig. figs.        | 0.1 mg l 1  |
| (E)Particulate Organic Carbon           | CPART            | mg l ً¦             | 3 sig. figs.        | 0.1 mg <sub>₁</sub> l⁻'                             |
| <sup>(K)</sup> Biological Oxygen Demand | BOD              | mg l ˈ              | 3 sig. figs.        | 1 mg l <sup>-</sup> '                               |
| Total Phosphorus                        | PTOT             | mg l '              | 3 sig. figs.        | 0.005 mg l  |
| Particulate Phosphorus                  | PPART            | mg l ້              | 3 sig. figs.        | 0.005 mg l 1  |
| Phosphate (soluble                      | PO4P             | mg l <sup>-</sup> ' | 3 sig. figs.        | 0.005 mg l <sup>-</sup> '                           |

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| reactive):PO <sub>4</sub> –P                   |       |                    |              |                                 |
|--|-------|--------------------|--------------|---------------------------------|
| Silicate: SiO <sub>2</sub>                     | SIO2  | mg l <sup>-1</sup> | 3 sig. figs. | 0.01mg l <sup>-1</sup>          |
| Sulphate: SO <sub>4</sub> –S                   | SO4S  | mg l <sup>-1</sup> | 3 sig. figs. | 0.005 mg l <sup>-1</sup>        |
| Sodium – dissolved                             | NADIS | mg l <sup>-1</sup> | 3 sig. figs. | 0.01mg l <sup>-1</sup>          |
| Sodium – total <sup>(8)</sup>                  | NATOT | mg l <sup>1</sup>  | 3 sig. figs. | $0.01 \text{mg} \text{ I}^{-1}$ |
| Potassium – dissolved                          | KDIS  | mg l <sup>-1</sup> | 3 sig. figs. | 0.01mg l <sup>-1</sup>          |
| Potassium – total <sup>(8)</sup>               | КТОТ  | $mg l^{-1}$        | 3 sig. figs. | $0.01 \text{mg}^{-1}$           |
| Calcium – dissolved                            | CADIS | $mg l^{-1}$        | 3 sig. figs. | $0.01 \text{mg}^{-1}$           |
| Calcium – total <sup>(8)</sup>                 | CATOT | $mg l^{-1}$        | 3 sig. figs. | $0.01 \text{mg}^{-1}$           |
| Magnesium – dissolved                          | MGDIS | mg l <sup>-1</sup> | 3 sig. figs. | 0.01mg l <sup>-1</sup>          |
| Magnesium – total <sup>(8)</sup>               | MGTOT | $mg l^{-1}$        | 3 sig. figs. | $0.01 \text{mg}^{-1}$           |
| Aluminium – total <sup>(8)</sup>               | ALTOT | µg l <sup>-1</sup> | 3 sig. figs. | 10 µg Ĭ <sup>1</sup>            |
| Aluminium – labile <sup>(9)</sup>              | ALLAB | ua l <sup>-1</sup> | 3 sia. fias. | 10 µg   <sup>-1</sup>           |
| <sup>(R)</sup> Tin – dissolved                 | SNDIS |                    | 3 sig. figs. | 1 µg   <sup>-1</sup>            |
| $^{(R)}$ Tin – total $^{(8)}$                  | SNTOT |                    | 3 sig. figs. | $1 \mu q l^{-1}$                |
| Manganese – dissolved                          | MNDIS |                    | 3 sia. fias. | 50 µg   <sup>-1</sup>           |
| Manganese – total <sup>(8)</sup>               | MNTOT |                    | 3 sig. figs. | 50 µg   <sup>-1</sup>           |
| Iron – dissolved                               | FEDIS | $\mu g l^{-1}$     | 3 sig. figs. | 50 $\mu g l^{-1}$               |
| Iron – total <sup>(8)</sup>                    | FETOT | ug l <sup>-1</sup> | 3 sig. figs. | 50 µg l <sup>-1</sup>           |
| <sup>(R)</sup> Vanadium – dissolved            | VDIS  | $\mu g l^{-1}$     | 3 sig. figs. | $4 \mu g l^{-1}$                |
| <sup>(R)</sup> Vanadium – total <sup>(8)</sup> | VTOT  | $\mu g l^{-1}$     | 3 sig. figs. | $4 \mu g l^{-1}$                |
| <sup>(R)</sup> Nickel – dissolved              | NIDIS | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g l^{-1}$                |
| <sup>(R)</sup> Nickel – total <sup>(8)</sup>   | NITOT | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g l^{-1}$                |
| <sup>(R)</sup> Mercury – dissolved             | HGDIS | $\mu g l^{-1}$     | 3 sig. figs. | $0.1 \mu g  l^{-1}$             |
| <sup>(R)</sup> Mercury – total <sup>(8)</sup>  | HGTOT | ug l <sup>-1</sup> | 3 sig. figs. | $0.1 \ \mu g \ l^{-1}$          |
| <sup>(R)</sup> Copper – dissolved              | CUDIS | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g I^{1}$                 |
| <sup>(R)</sup> Copper – total <sup>(8)</sup>   | CUTOT | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g l^{-1}$                |
| <sup>(R)</sup> Zinc – dissolved                | ZNDIS | $\mu g l^{-1}$     | 3 sig. figs. | $2 \mu g l^{-1}$                |
| <sup>(R)</sup> Zinc – total <sup>(8)</sup>     | ZNTOT | $\mu g l^{-1}$     | 3 sig. figs. | $2 \mu g l^{-1}$                |
| <sup>(R)</sup> Cadmium – dissolved             | CDDIS | $\mu g l^{-1}$     | 3 sig. figs. | $0.1 \mu g  l^{-1}$             |
| <sup>(R)</sup> Cadmium – total <sup>(8)</sup>  | CDTOT | $\mu g l^{-1}$     | 3 sig. figs. | $0.1 \ \mu g \ l^{-1}$          |
| Lead – dissolved                               | PBDIS | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g I^{-1}$                |
| Lead – total <sup>(8)</sup>                    | PBTOT | $\mu g l^{-1}$     | 3 sig. figs. | $1 \mu g l^{-1}$                |
| Arsenic – total <sup>(8)</sup>                 | ASTOT | µg l <sup>-1</sup> | 3 sig. figs. | 10 µg l <sup>-1</sup>           |
| For each determinand.                          |       |                    |              |                                 |
| Laboratory Code <sup>(10)</sup>                |       |                    |              |                                 |
| Limit of Detection Code <sup>(11)</sup>        |       | character code (<) |              |                                 |
| Analysis Date                                  |       |                    |              |                                 |

### Notes

The prefixes (R) and (L) before the determinand names above indicate that the analysis is to be performed for rivers only or lakes only, respectively. Where no prefix occurs, the determinand applies to both rivers and lakes.

- <sup>(1)</sup> These codes should be used within the analytical dataset exactly as given in the Table above. Any additional determinands to be included may be allocated codes by agreement with the ECN Data Manager.
- (2) These limits of detection are those recommended by the ECN Statistical and Technical Advisory Group as necessary for the detection of environmental change in the listed determinands.
- $^{(3)}$  pH should be measured (or compensated for) at 20°C.
- <sup>(4)</sup> Suspended Solids: Dry weight should be determined at 105°C  $\pm$ 5° and ash-free dry weight should be determined at 500°C  $\pm$ 20°.
- <sup>(5)</sup> Turbidity and Secchi disc are alternatives to Suspended Solids.
- <sup>(6)</sup> Lake profiles: Temperature and Dissolved Oxygen should be recorded at depths considered appropriate to the profile of each particular lake. Information about recording depths should be provided and a numeric suffix to the TEMP and DISOX determinand codes (eg TEMP1, TEMP2, etc) used to identify each depth position.
   <sup>(7)</sup> Conductivity about the provided has a compared for a segment of factors.
- <sup>(7)</sup> Conductivity should be measured (or compensated for) at 25 °C.
- <sup>(8)</sup> Please note that 'total' implies analysis of the unfiltered sample. If the laboratory routinely filters samples and can only provide 'total dissolved', then the code xxTOTD,

where xx is the symbol for the metal, should be used. For example: ALTOTD for total dissolved Aluminium.

- <sup>(9)</sup> Labile Aluminium should only be measured at sites where there is a record of the pH ever having been less than 5.5.
- <sup>(10)</sup> The Laboratory Code provides the link with the laboratory methods information (see below) to be supplied by freshwater organisations for each of their laboratories. The code should incorporate the organisation acronym where possible and should be agreed the ECN Data Manager.
- (11) Where the value of a particular determinand falls below the limit of detection (LOD) for the method, the value of the determinand should be given as the LOD value, and the LOD code set to the character <. Where the value is at or above the detection limit this code should be left null.

### Laboratory methods information

Details of analytical methods used by each laboratory involved in ECN should be submitted to the ECN Data Manager. The information is stored in the ECN metadatabase and linked with the data *via* the Laboratory Code, Determinand Code and Date range. When any details change, a new record should be submitted by the laboratory and will be added to the database. Aspects of the analysis such as instrument maintenance, calibration, drift, and training of staff will be under the control of the laboratory. The text format for submitting methods information is illustrated below, using nitrate as an example:

| Organisation                   | NERC   |
|--------------------------------|--|
| Laboratory                     | ITE, Merlewood   |
| Laboratory Code <sup>(1)</sup> | ITE-ME   |
| Substance determined           | Nitrate  |
| Determinand code               | NO3N   |
| Basis of the method            | Chemically Suppressed Ion Chromatography   |
| Types of sample                | Stream water (FWC)   |
| Typical concentrations         | FWC: 0.50  |
| Volume for analysis            | 10 ml  |
| Calibration range              | 0.01 to 10 mg l <sup>-1</sup> – slight deviation from linearity                                |
|                                | corrected for by using 3rd-order regression.   |
| Method of measurement          | Peak area using integration/data system.   |
| Results reported               | 3 sig figs as N(mg l <sup>-1</sup> )   |
| Detection limit <sup>(2)</sup> | 0.01 mg l <sup>-1</sup>  |
| Within batch std. devn.        | -  |
| (mid range) <sup>(3)</sup>     | 2% rsd   |
| Interferences                  | None   |
| Internal QC measure            | CUSUM quality control chart  |
| Accuracy measure               | AQUACHECK  |
| Method                         | Dionex 2002i ion chromatograph with 50 µl injection  |
|                                | loop, auto-sampler and sample load pump.   |
|                                | Columns –AG4A–SC & AS4A operated with an anion   |
|                                | micro-membrane suppressor and using an eluent mixture  |
|                                | of 0.15 mM NaHCO <sub>3</sub> /2mM Na <sub>2</sub> CO <sub>3</sub> at 1.8 ml min <sup>-1</sup> |
| Reference                      | Merlewood Lab method 3.7.5   |
| Method used from date          | 11-NOV-1992 (the date on which analysis commenced  |
|                                | using this method)   |
| Method used until date         | (the final date on which this analysis was used - leave  |
|                                | blank if current)  |

### Notes

- <sup>(1)</sup> The Laboratory Code should incorporate the organisation acronym where possible. A code should be agreed with the ECN Data Manager.
- <sup>(2)</sup> The detection limit is defined as 4.65 within-batch standard deviation of the blank or a solution with a concentration close to the blank when no signal is detectable from the blank (n = 10).
- <sup>(3)</sup> A within-batch standard deviation in excess of 5% is unlikely to be acceptable.

# Core measurement: Freshwater physical and chemical measurements – automatic sampling (FWA protocol)

|                           |                | Precision of |
|---------------------------|----------------|--------------|
| Variable                  | Units          | recording    |
| Site Identification Code  |                |              |
| Core Measurement Code     |                |              |
| Location Code             |                |              |
| Recording (Sampling) date |                |              |
| Recording (Sampling) time | GMT 24-h clock | 1 min        |
| pH (average)              | pH units       | 0.1          |
| Temperature (average)     | °C             | 0.1          |
| Conductivity (average)    | µS cm⁻¹        | 0.1          |
| Turbidity                 | NTU            | 1            |

The following variables are recorded automatically every hour: